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## **SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)**

Chapter 15 water and aqueous systems. Objectives... Properties of water. Solute, solvent, solution. Electrolytes. Suspension, colloid, emulsion. Many unique and important properties of water result from hydrogen bonding. high surface tension. low vapor pressure. Surface Tension. In other liquids, hydrogen bonds form between molecules on all sides of the liquid (balanced) In water, hydrogen ...

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use these activities to yourself study the vocabulary and major concepts presented in this chapter

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molecule are polar covalent bonds. The covalent bonds are polar because the oxygen atom has a greater electronegativity than the hydrogen atoms. The bonds between adjacent water molecules are called hydrogen bonds.

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